



materials design

A comprehensive environment for
property prediction and force field
development

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Agenda

- MEDEA software platform
- Properties
- Examples:
 - Vapor Pressure and VLE
 - Azeotropes
 - Phase Diagram
 - Thermal conductivity
 - Viscosity
- Intrinsic Issues in Calculations
- Challenges for industrial R&D
 - Time Pressure
 - Multiple Projects
 - Restricted Environment
- Assessment

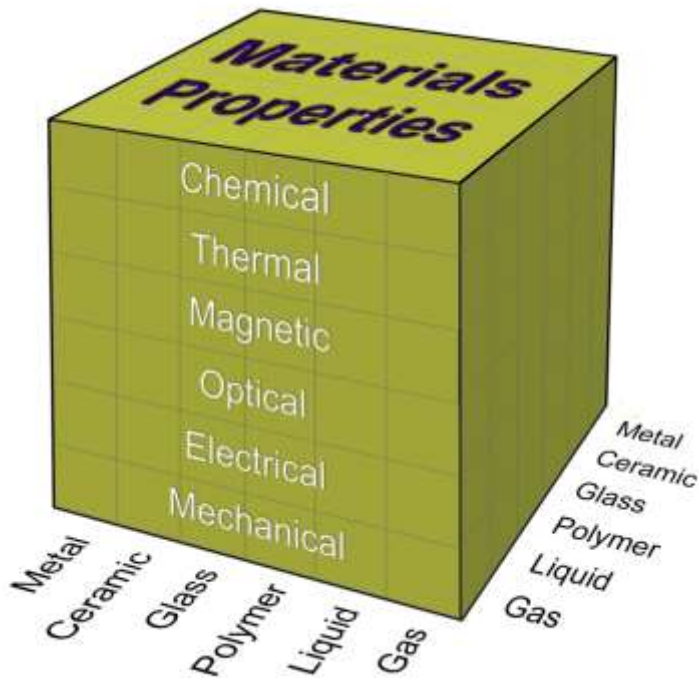


MEDEA SOFTWARE PLATFORM



Purpose

Predict and understand materials properties

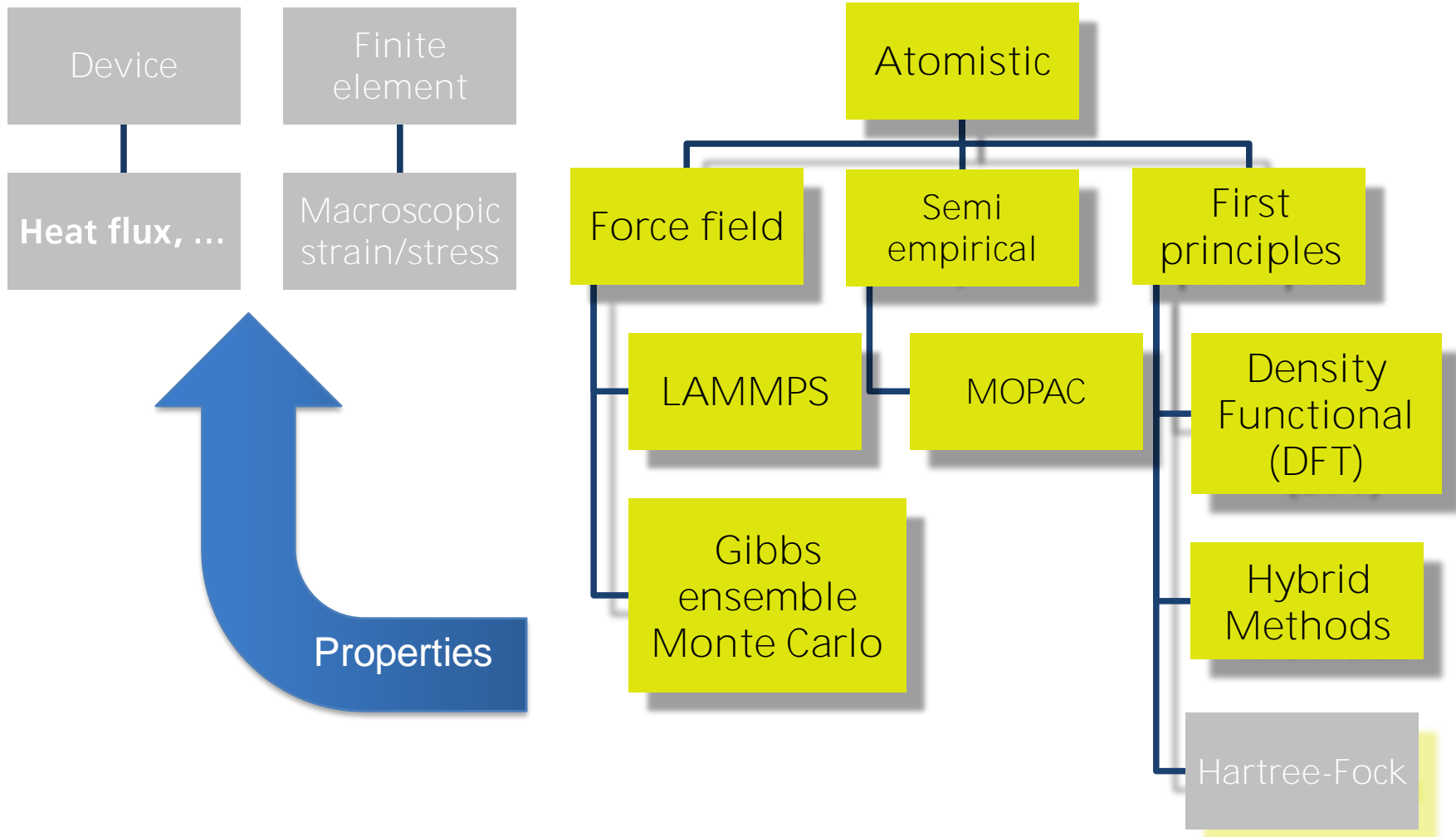


Approaches:

- ▶ Ab initio electronic structure calculations
- ▶ Forcefield simulations
- ▶ Statistical mechanics
- ▶ Analytical theory
- ▶ Empirical correlations
- ▶ Experimental data of existing materials (databases) as reference



Materials modeling: MedeA capabilities



MedeA's Three Tier Architecture



User Interface

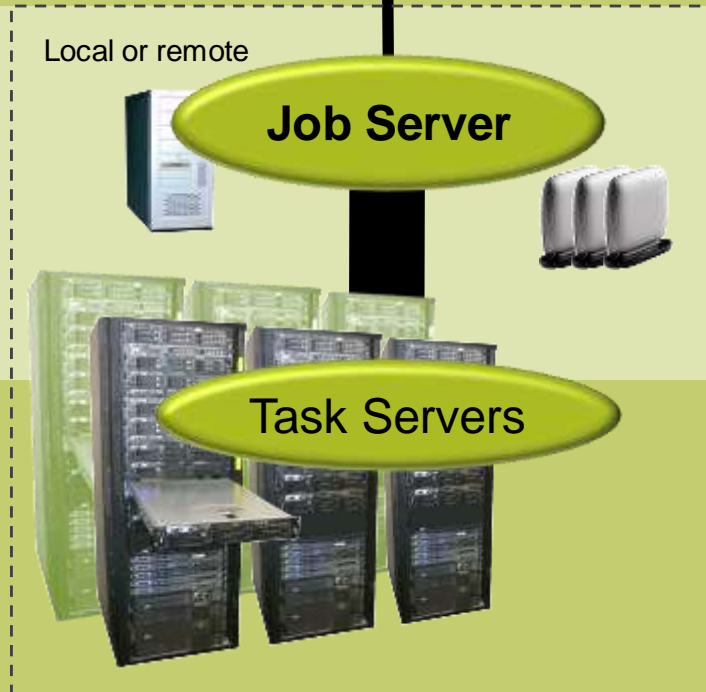


Model creation
and analysis of results

Local databases
(expt. and computed)



Job control and
job databases



Computations

VASP GIBBS LAMMPS MOPAC
Integration with

Queuing System: PBS, LSF, ...



PROPERTIES



Forcefield Methods in MedeA

- Focus on materials properties including:
 - Density and structure
 - Vapor pressure
 - Solubility/miscibility
 - Liquid-vapor phase diagrams, critical points
 - Thermal conductivity
 - Viscosity
 - Mechanical properties
 - Diffusion
- Selection, editing, and control of forcefields
- Automated atom-typing and forcefield assignment
- Flow-chart control of computational stages
- Full use of parallel architectures
- Property-oriented analysis tools
- Estimation of (statistical) error bars

Simulation Flowchart

Calculation | Preview Input

Reading and writing flowcharts

Open ... Save ...

From job: ... Open

Add stages

Initialization

Start

Initialize LAMMPS

Set variables

Control

For all

Building

Set cell

Minimization

Minimize

Dynamics

Initialize velocities

nVE ensemble

nVT ensemble

nPT ensemble

Thermal Conductivity

Viscosity

Custom

Custom code

LAMMPS Flowcharts

```
007: (C3 O3 H8)250 (P1) - glycerol_250 0 1 1 x 1 0 1 x 1 1 0 -
009: (C3 O3 H8)125 (P1) - glycerol_125 (LAMMPS)
005: (C3 O3 H8)125 (P1) - glycerol_rho=1.154 Green-Kubo VI
004: (C3 O3 H8)125 (P1) - gly_125_rho1154 (LAMMPS)
011: (O82 O2 H9)4 (P1) - LAMMPS_dynamics 500ps nVT_50
010: (O82 O2 H9)4 (P1) - H2O = CNT_minimization
```

OK Cancel

Title: (C3 O3 H8)125 (P1) - glycerol_125 (LAMMPS)

Run Close Write input file

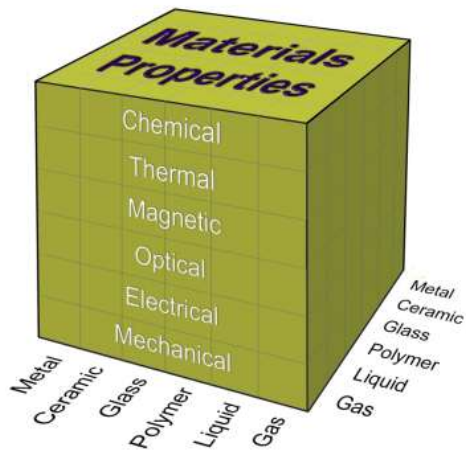


Forcefield Control

- **Automated atom type assignment** using the templates section of Materials Design's forcefield file *.frc
- **Wildcards:** major simplification in the angle and torsion terms – critical for more complex simulations with GIBBS and LAMMPS
- **Atom type equivalences** for nonbonds, bonds, angles, torsions, etc.
- **Versioning:** each parameter has its own version, so updates do not remove older parameters but override them
- **“Include” capability:** a user can modify a forcefield by including the original, adding parameters and, by using version numbers, override parameters in the original in a well organized and controlled manner



Classes of Materials



- Gases and liquids including high pressure and temperature
- Organic materials ranging from small molecules to polymers
- Inorganic materials (e.g. crystalline, amorphous materials, glasses, molten materials)
- Semiconductor materials (Si, Ge, III-V, ...)
- Metals and alloys
- Nanostructures
- Interfaces



Computed Materials Properties

Structural properties

- Molecular structures
- Crystal structures
- Surface structures
- Structure around defects
- Adsorption geometries
- Structures of interfaces
- Liquids and amorphous systems

Thermo-Mechanical properties

- Elastic moduli
- Speed of sound
- Vibrational properties
- Thermal expansion coefficients
- Fracture

Thermodynamic properties

- ΔU , ΔH , ΔS , ΔG , heat capacity
- Binding energies
- Solubility
- Melting temperature
- Vapor pressure
- Miscibility
- Phase diagrams
- Surface tension

with forcefields

Chemical properties

- Chemical reaction rates in gases and condensed phases
- Reactivity on surfaces
- Solid-solid reactions
- Pressure-induced reactions
- Photochemical reactions

Transport properties

- Mass diffusion coefficient
- Permeability
- Thermal conductivity
- Viscosity

with
forcefields

Electronic, optical, and magnetic properties

- Electron density distribution - electrical moments
- Polarizabilities, hyperpolarizabilities
- Optical spectra
- Dielectric properties
- Piezoelectric properties
- Electrostatic potential
- Spin density distribution, magnetic moments
- Energy band structure - metal, semiconductor, insulator, superconductor
- Band gaps, band offsets at hetero-junctions
- Ionization energies and electron affinities
- Work function



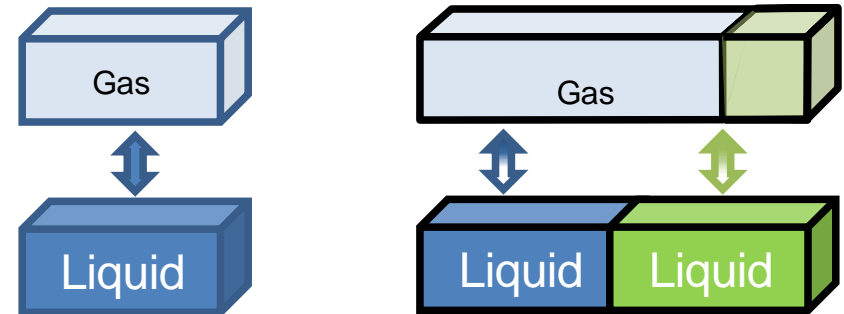
EXAMPLES



Carbon Dioxide

- Vapor pressure
- Density gas-liquid
- Ethane-CO₂ azeotrope
- CO₂-SO₂ phase diagram
(Lachet et al., 2009)
- Viscosity
- Thermal conductivity

Gibbs ensemble Monte Carlo
Biased configurational averages

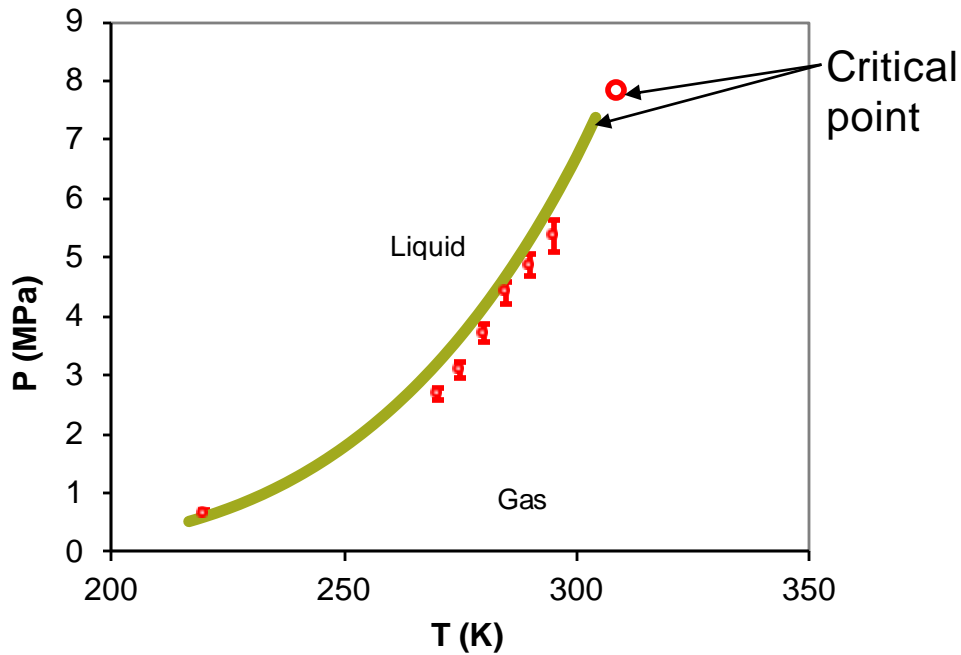


LAMMPS

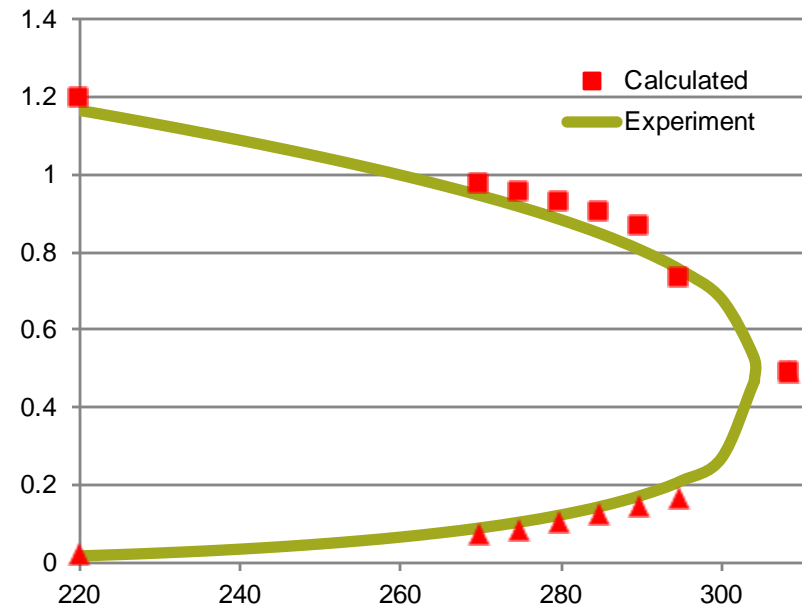


Results for CO₂

Vapor pressure

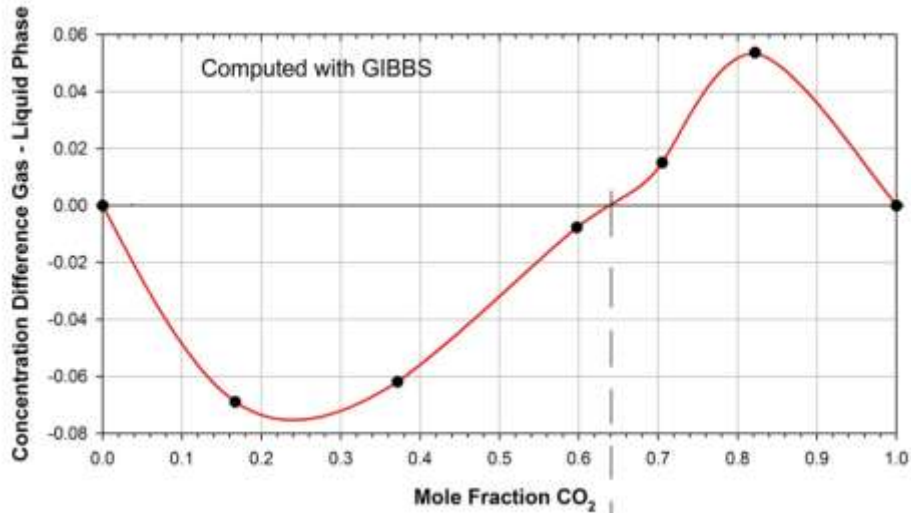


Density of liquid and gas phase





Ethane-CO₂ Azeotrope



C₂H₆ - CO₂

Standard GIBBS AUA potentials

Computed concentration of CO₂ of azeotrope at 34 bar:

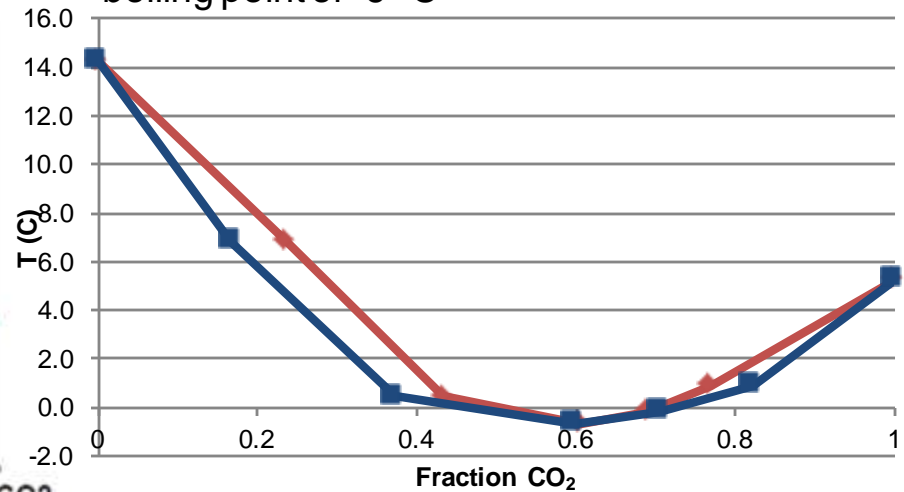
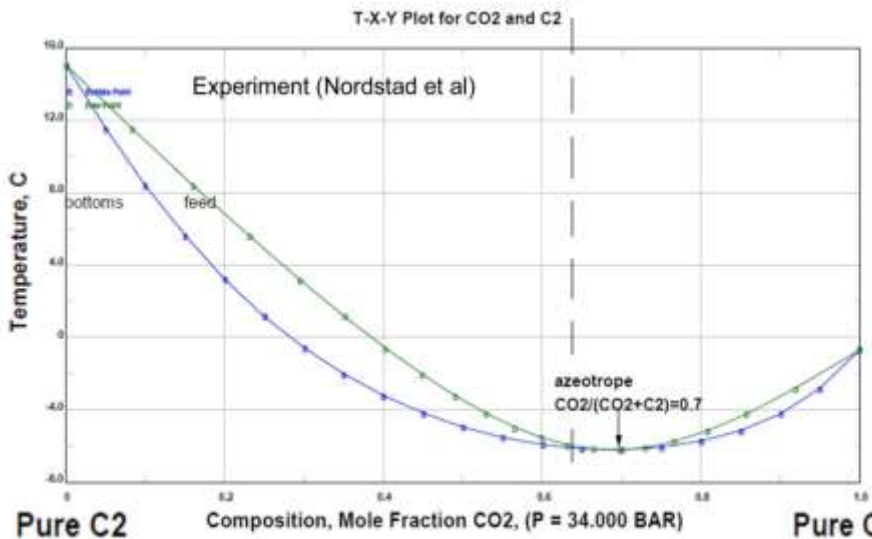
[CO₂]=0.64

boiling point of -0.6 °C

Experiment (Nordstad et al.):

[CO₂]=0.7

boiling point of -6 °C





CO₂ – SO₂ Phase Diagram

Energy Procedia 1 (2009) 1641–1647

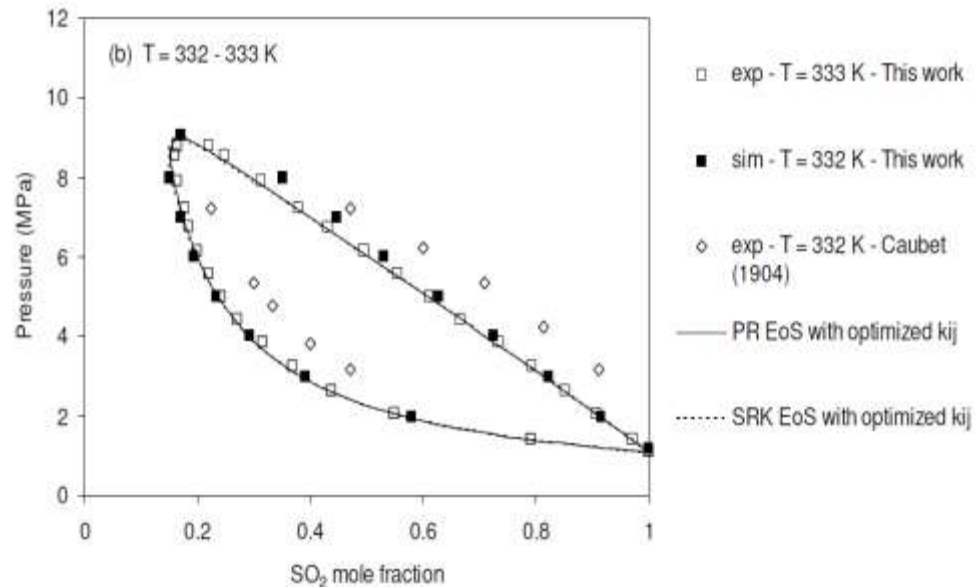
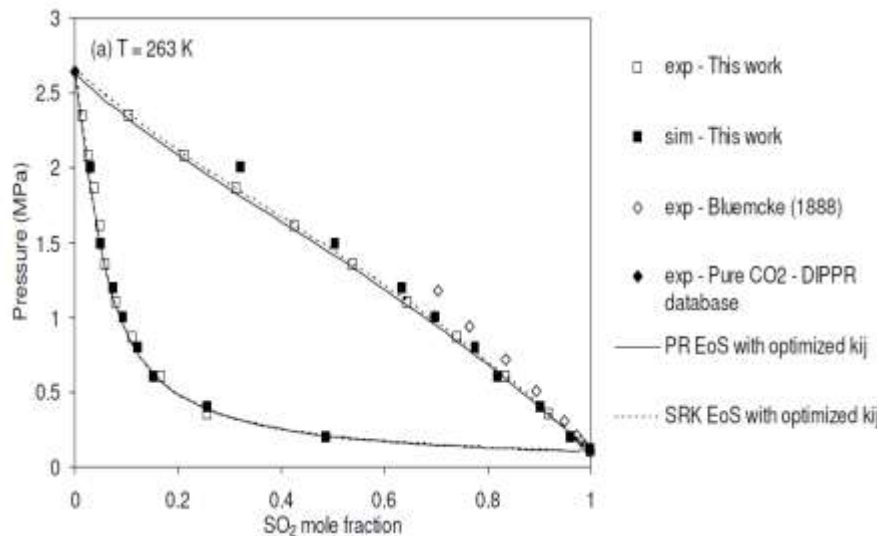
Thermodynamic behavior of the CO₂ + SO₂ mixture: experimental and Monte Carlo simulation studies

Véronique Lachet^{a,*}, Theodorus de Bruin^a, Philippe Ungerer^a, Christophe Coquelet^c,
Alain Valtz^c, Vladimir Hasanov^b, Frederick Lockwood^b, Dominique Richon^c

^aIFP, 1-4 avenue de Bois Préau, 92852 Rueil-Malmaison, France

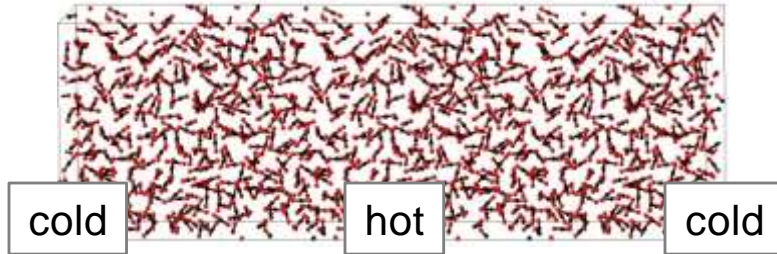
^bAir Liquide, 1 chemin de la Porte des Loges, Les Loges-en-Josas BP 126, 78354 Jouy-en-Josas, France

^cMines Paris-Tech, CEP/TEP, Centre Energétique et Procédés CNRS FRE 2861, 35 rue Saint Honoré, 77305 Fontainebleau, France

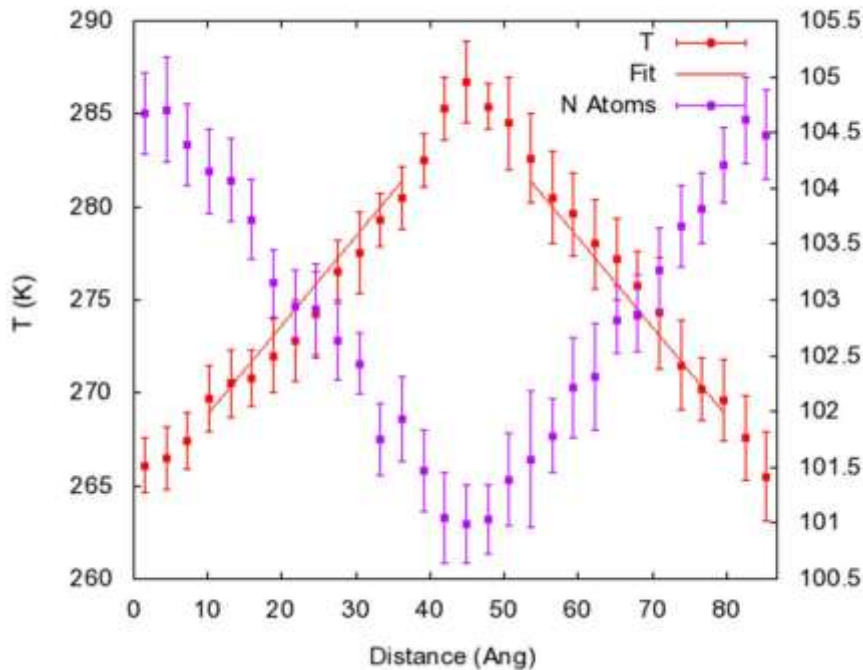




Thermal Conductivity



Temperature Profile for Stage IV



CO₂

$P = 500$ bar

$V = 2.89 \times 2.89 \times 8.68$ nm³

$T = 298.2$ K

$\rho = 1.0341$ g/cm³

5 ns molecular dynamics

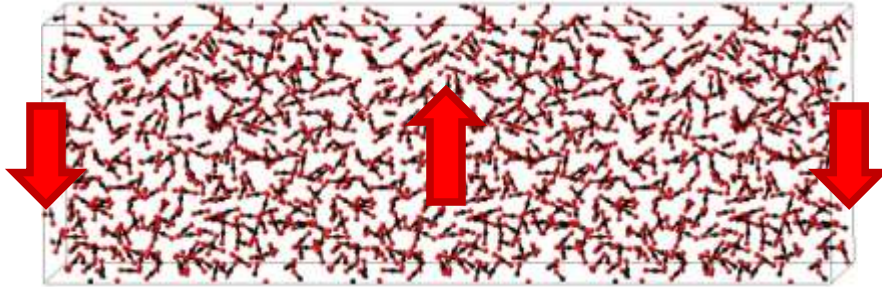
$\lambda_{comp} = 0.141 \pm 0.017$ W m⁻¹ K⁻¹

$\lambda_{expt.} = 0.1385$ W m⁻¹ K⁻¹

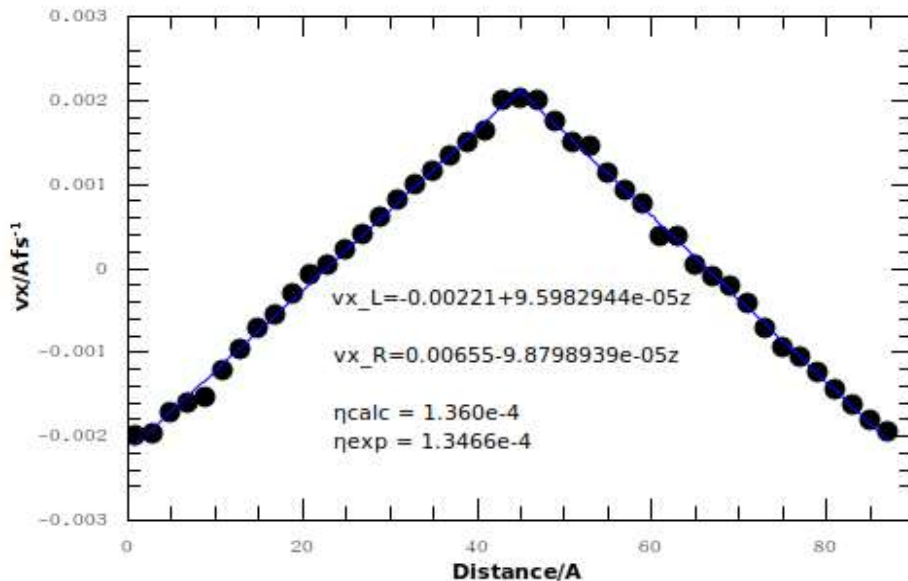
Reverse non-equilibrium molecular dynamics
MedeA-LAMMPS
COMPASS forcefield



Viscosity



CO₂ - T = 298.2K - ρ = 1.0341 - P = 494 atm
(3087 atoms)



CO₂

P = 494 bar

V = 2.89 x 2.89 x 8.68 nm³

T = 298.2 K

ρ = 1.0341 g/cm³

0.1 ns molecular dynamics

$\eta_{comp} = 1.360 \times 10^{-4}$ Pa s

$\eta_{expt.} = 1.3466 \times 10^{-4}$ Pa s (NIST)

Reverse non-equilibrium molecular dynamics

MedeA-LAMMPS

COMPASS forcefield



INTRINSIC ISSUES WITH CALCULATIONS

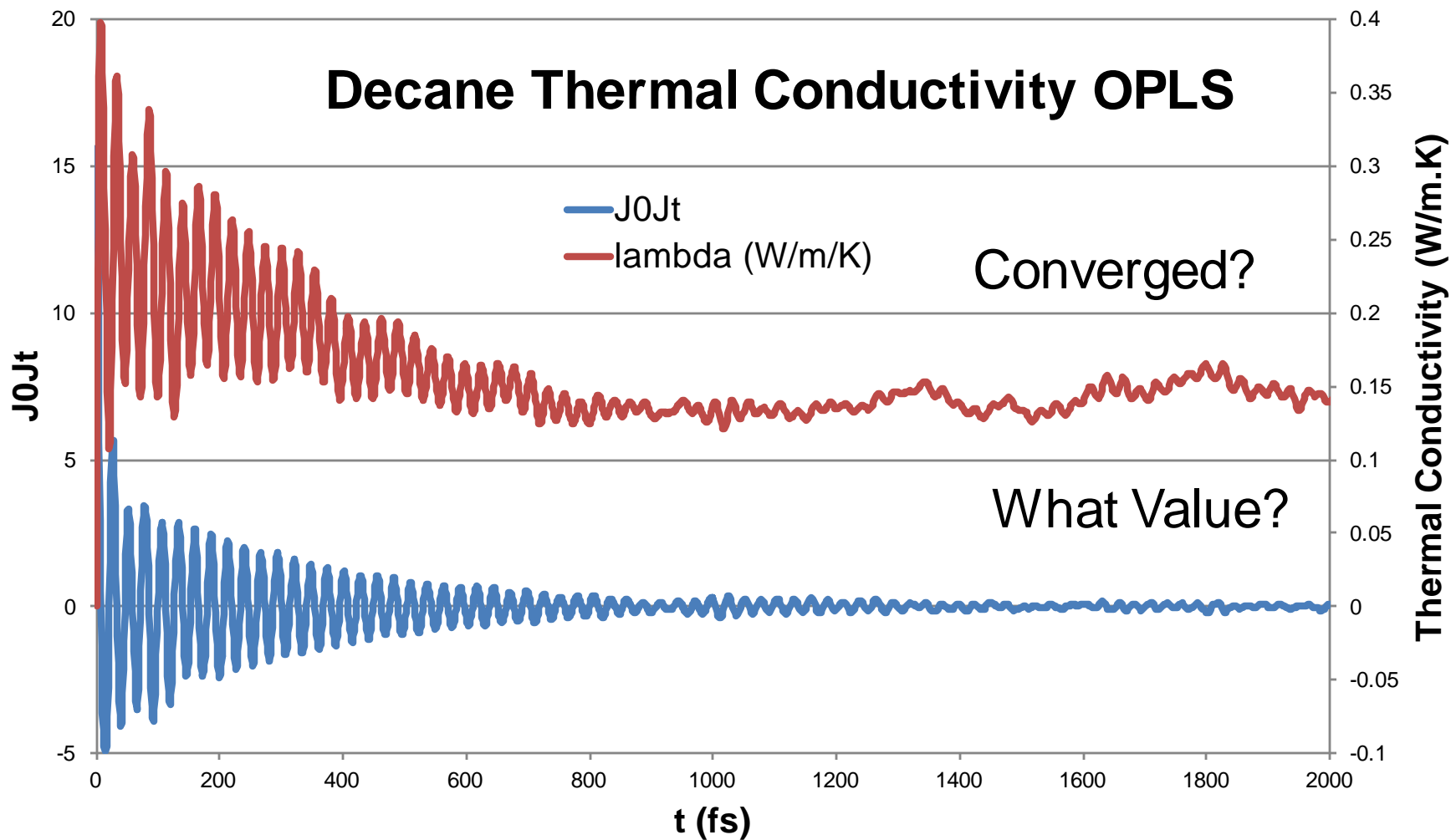


Forcefield Accuracy

Alkanes and Alcohols				
Liquid	Temperature	Density (expt)	Density LAMMPS/OPLSAA	Error
	(K)	(g/cm ³)	(g/cm ³)	
Butane	273	0.6013	0.6010 ± .0026	-0.05%
Isobutane	273.2	0.58052	0.6046 ± .0039	4.15%
Isobutane	298.2	0.55059	0.5709 ± .0056	3.69%
Pentane	298.2	0.62074	0.6156 ± .0032	-0.83%
Isopentane	298.2	0.61516	0.6261 ± .0032	1.78%
Neopentane	298.2	0.58435	0.6276 ± .0036	7.41%
Methanol	298.2	0.786	0.7797 ± .0021	-0.80%
Ethanol	298.2	0.78509	0.7955 ± .0021	1.30%
Isopropanol	300	0.7795	0.8088 ± .0036	3.76%
1,2-butanediol	373.2	0.9394	0.9591 ± .0037	2.10%
Glycerol	373.2	1.209	1.1893	-1.60%



Analyzing the Results





Which ARE the Right Results?

Decane at 480 K and 178 atm

SHAKE _n ?	COMPASS	OPLS
No	0.19	0.17
Yes	0.16	0.13
(experiment)	0.10015	W/m.K

SHAKEing C-H bonds and H-C-H.

Even at 480 K C-H stretch and the H-C-H and C-C-H bends will not be active. What about C-C-C?

How does this affect calculated thermal conductivity?



CHALLENGES FOR INDUSTRIAL R&D



Challenges for Industrial R&D

– Immediate Proof of Concept

- Can we calculate “x” for “y”?
 - Very short timeframe – hours or perhaps a day or two
 - What – exactly – is “y” in terms of an atomic model?
 - Need to answer “What is the accuracy of the result?”
- Wide range of properties and systems – no one tool fits all

– Projects

- Tight schedules
- Aren’t simple, by definition!
- Usually require a large number of calculations – 100’s
- One of many simultaneous projects

– Proprietary

- Not easy to find related work without revealing topic
- How to get help with problems with a specific system?



Challenges of Industrial Environment

- **Lack of control over computing resources**
 - Shared resources
 - IT restrictions
 - Type of hardware
 - Network access
 - Security
 - Changing passwords monthly
- **Software Environment**
 - Must be robust, robust, robust!
 - Handling of a large number (>10,000/yr) of a wide range of calculations
 - Well automated, including analysis
 - Checking for errors
 - Providing a history
 - Error bars

As Anne Chaka of NIST comments: “*We give away the data for free – but we charge for the error bars!*”



From code to property

- Programmer's Mind

- *#kpoints*
- Optimize program
 - Platform specific
 - Special cases:
 - Non-magnetic
 - Independent blocks
 - Block/group per node
- Expert in code
- Full Focus

- Engineers Mind

- Similar *k-spacing*
- General approach
 - Platform independent
 - Keep procedure
 - Change molecule
 - Change process property
 - Use more computing power
- Expert
- One of many tools



ASSESSMENT



Assessment

- Computer simulations have become a viable source of many industrially important thermodynamic property data providing
 - Consistent and complete dataset
 - Resolution of conflicting experimental data
 - Extrapolation to regions where experiments are too difficult, costly, time consuming, or dangerous
 - Understanding on deep level
- We are still in an early stage; accuracy and cost-efficiency will accelerate deployment
- The key bottleneck is “*cognitive access*”,
 - scientists and engineers who understand industrial needs
 - have the skills to formulate meaningful simulations
 - and control the error bars



I Have a Dream

- Our technological world faces enormous challenges
- The solutions – for better or worse – must rely on technology
- We must always do better!
- Atomistic modeling can play an important, growing role



Two Routes Forward

- **Open Source/Content**
- **Commercial Companies**



This is Work!

- Integrating these methods and tools is much more work than developing them in the first place – 10x?
- This is science, not software
- The intellectual investment is enormous
- Think chips and computers, not Linux or Wikipedia



Example: Forcefields

- Forcefields are highly **interconnected**:
new parameters must work with and extend preceding parameters
- Require large effort to **validate**
- Require **consistency**
- Are a dynamic **compromise** between coverage and accuracy
- Typically the work of a single group, headed by one person



Organic Forcefields

- No publicly available accurate organic forcefields!
 - OPLS ... sort of, but quite uneven
 - MM4 ... ?
 - COMPASS ... commercial
- No longer actively developed
 - Last major efforts were at Accelrys > 10 years ago
 - Allinger and MM4 8-10 years ago
- Academic forcefield work focussing on new frontiers
 - Reactive forcefields
 - Polarizable forcefields
 - Etc.
- Where do the next generation of forcefields come from?



What Industry Needs

- Modeling tools that are so useful that they cannot be ignored!
- Robust, competitive commercial products
- Leading edge academic research moving into products over time
- Intel and AMD to compete and compete and...
- National Labs to demonstrate large scale use, also advance high computing